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Note

# Evacuated blood collecting tubes as containers for taking and storing gas samples

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Gas samples for analysis may be collected in containers which are variations of either tubing with a valve at each end or a syringe with gas-tight seals and a valve at the intake end. These are excellent for taking and delivering one sample at a time but are costly and inconvenient when taking large numbers of samples at remote locations. This report describes an evaluation of evacuated, rubber-stoppered glass tubes manufactured for blood collection as the vehicle for collection and storage of gas samples for gas chromatographic analysis.

Gas samples were analyzed by gas chromatography<sup>1,2</sup> using a Hewlett-Packard 7620 gas chromatograph fitted with thermal conductivity detector, liquid-CO<sub>2</sub> cooling and integrator. The chromatograph was fitted with dual 6 ft.  $\times \frac{1}{4}$  in. stainless-steel columns of a Porapak Q (80-400 mesh). The temperature program of the oven was: 4 min isothermal at -60°; temperature increase 30°/min to 90°, isothermal 2 min at 90°; temperature increase 30°/min to 170° and 4 min isothermal at 170°. Helium was used at the carrier gas. Gas samples (500 µl) were injected into the chromatograph using 500-µl gas-tight syringes fitted with closure valves between syringe and needle.

Gases used in this study were: He; CH<sub>4</sub> (natural gas, >99% pure CH<sub>4</sub>) from laboratory gas cocks; 3.01% CO<sub>2</sub>, balance N<sub>2</sub>; and 11.91% CO<sub>2</sub>, balance O<sub>2</sub>. The latter two are normally used for calibration of blood gas analyzers<sup>\*</sup>. Mixtures of these gases were prepared by downward displacement of water. Control gas samples at CO<sub>2</sub>-N<sub>2</sub>or CO<sub>2</sub>-O<sub>2</sub> mixtures and CH<sub>4</sub> were taken by direct sampling with the injection syringe from the wet electrode compartments of a Radiometer BEU-1 blood electrode unit.

Evacuated blood collecting tubes (Venoject<sup>®</sup>)<sup>\*\*</sup> without anticoagulant were used. All tubes were from one manufacturing lot and were taken from laboratory stock. The tubes were filled with gas mixture using the standard tube holder fitted with a bleeding needle inserted into a contained gas mixture through a rubber injection stopper. The evacuated tube was pressed onto the adapter needle and gas was allowed to fill the tube to atmospheric pressure.

<sup>\*</sup> Purchased from Gas Dynamics. Scarborough, Ontario, Canada.

<sup>\*\*</sup> Manufactured by Jintan Terumo, Tokyo, Japan; distributed by Standard Hospital Supply, Calgary, Canada. Other makes include Vacutainor, Becton Dickinson, Mississauga, Ont., Canada; Monoject, Sherwood Industries, St. Louis, Mo., U.S.A.

| Sample and treatment   | No. 6 | No. of Area of 500-ul peaks (µvolt/sec) ± S.D. (CV) | (µvolt/sec) ±                   | : S.D. (CV)               |                                      | Gas comp                 | Gas composition (area corrected. %) |                      | + 5.D.                        |
|--|-------|---|---------------------------------|---------------------------|--------------------------------------|--------------------------|-------------------------------------|----------------------|-------------------------------|
|  | ples  | N <sub>1</sub> 0 <sub>2</sub>                       |                                 | CH4                       | CO1                                  | N1                       | 01                                  |                      | co1                           |
| Gas in blood tubes,<br>Hè makeup   | 39    | $120000 \pm 3340 36$<br>(2.78)                      | $36700 \pm 2000$                | <b></b>                   | 1                                    | <u>78.0</u> ± 2          | 2.2 22.0 ± 1.18                     | 1                    |                               |
| CH4, free ** (natural gas)   | S     | 9400 ± 850<br>(9.06)                                |                                 | 495000 ± 12500            | [                                    | 1.2 ± 0                  | 0.1                                 | 98.8 ± 2.5           | I                             |
| From blood tubes   | ŝ     | 1040 37   |                                 | (22.2) 1310 373000 ± 4270 | ł                                    | 17.0 ± 0                 | <b>0.1</b> 4.6 ± 0.2                | <b>78.4</b> ± 0.9    | I                             |
| Blank corrected <sup>*</sup><br>3.01 % CO <sub>2</sub> -N <sub>1</sub> , free ** | 10    | 0200  | 300                             | (1.14)<br>373000<br>      | 23000 ± 1336                         | 1/10.9<br>97.0 ± 1.      | 1.6                                 | 1.09<br>             | 2.9 + 0.2                     |
| From blood tubes   | 10    | (1.02)<br>(91700 ± 20700 369<br>(3.00)              | 36500 ± 1820<br>24 605          | ł                         | (5.8)<br>17560 <u>+</u> 927          | <b>93.5</b> ± <b>2.8</b> | ·8 4.5 ± 0.2                        | I                    | 2.0 ± 0.1                     |
| Blank corrected *<br>11.91 % CO-O, free **                                       | 10    |   | (4.56)<br>-200<br>619600 ± 9700 | 1 1                       | (1.72)<br>17560<br>102800 $\pm$ 1765 | 97.5<br>                 | —0.1<br>87.1 土 1.4                  | 11                   | 2.6<br>12.9 ± 0.2             |
| From blood tubes   | 10    | 4600  | (10.1)                          | 1                         | (1.72)<br>80250 $\pm$ 1530           | 17.0 ± 0.7               | 7 72.8 ± 1.4                        | ł                    | 10.1 ± 0.2                    |
| Blank corrected *<br>Mixed natural and and                                       | 3     | 0006-   | (1.98)<br>479300                | I                         | (1.91)<br>80252                      | ł                        |                                     |                      | 12.2                          |
| $11.91\% CO_{2}-O_{2}$   | ŧ     |   |                                 |                           |                                      | 21.1 ± 0.7               | 3                                   | $38.2 \pm 0.4$       | $3.9 \pm 0.1$                 |
| Stored at 20° for 7 days   | 8     |   |                                 |                           |                                      | $20.6 \pm 0.8$           | (c2.2)<br>8 37.3 ± 0.5              | (1.07)<br>38.5 + 0.5 | (3.9)<br>$3.8 \pm 0.1$        |
| Stored at 20° for 14 days  | 10    |   |                                 |                           |                                      | (4.1)<br>42.6 + 18.4     |                                     | (1.36)               | (1.84)                        |
| Stored at -20° for 14 days 7   | 2     |   |                                 |                           |                                      | (43.2)<br>$20.7 \pm 0.5$ |                                     | (51.9)<br>38.9 土 0.4 | (46.1)<br>(46.1)<br>3.6 ± 0.1 |

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Test gas samples were drawn into blood collecting tubes from a constantly flowing gas stream flowing through a pair of bubble jars. Samples were drawn from the first bubble jar. The second jar provided a seal to prevent aspiration of air into the system during the sudden filling of the evacuated tubes.

Calibration curves for N<sub>2</sub>, O<sub>2</sub>, CO<sub>2</sub>, and CH<sub>4</sub> were prepared by analyzing 500-, 400-, 300-, 200-, 100- and 50- $\mu$ l samples of air, 11.91 % CO<sub>2</sub>-O<sub>2</sub> and CH<sub>4</sub>. The response of the instrument was linear between 4 and 12 % CO<sub>2</sub> and linear for N<sub>2</sub>, O<sub>2</sub> alone, O<sub>2</sub> in presence of N<sub>2</sub><sup>\*</sup>, and CH<sub>4</sub> between 1 and 70-100% composition. The relative responses of the detector relative to N<sub>2</sub> was calculated from the slopes to be: 1.1 for O<sub>2</sub> alone; 0.92 for O<sub>2</sub> in the presence of N<sub>2</sub><sup>\*</sup>; 1.55 for CH<sub>4</sub>; and 0.82 for CO<sub>2</sub>.

The reproducibility of the gas analysis itself was tested by analyzing 44 samples of laboratory air taken over a 7-day period. The area values for N<sub>2</sub> and O<sub>2</sub> were for N<sub>2</sub> and O<sub>2</sub> were 596200  $\pm$  11700 and 172000  $\pm$  2450, respectively, with coefficients of variation (CV) of 1.95 and 1.43. These correspond to air composition values of 76.1  $\pm$  1.5% N<sub>2</sub> and 20.9  $\pm$  0.3% O<sub>2</sub>.

The reproducibility of blank values (*i.e.*, the residual air in the evacuated blood tubes) was tested by filling tubes with wet He and analyzing the contained gases<sup>\*\*</sup>. The CV of  $N_2$  and  $O_2$  for the 39 tubes were 2.78 and 5.39, respectively (Table I).

Analysis of the CH<sub>4</sub> sampled from the BEU-1 showed the presence of small proportions of N<sub>2</sub> (1.91% of CH<sub>4</sub> integrated peak area) which had a CV of 9.06% and CH<sub>4</sub> with a CV of 2.52%. When CH<sub>4</sub> was sampled using blood tubes, the N<sub>2</sub>, O<sub>2</sub> and CH<sub>4</sub> measured (Table I) had CV values of 0.83, 0.35 and 1.14%, respectively. The known gas mixtures were analyzed as controls on 10 different days during the course of the trials described. The CV for the analysis of 3.01%-N<sub>2</sub> mixture were 1.62 and 5.81 for N<sub>2</sub> and CO<sub>2</sub>, respectively and for the 11.91% CO<sub>2</sub>-O<sub>2</sub> mixture were 1.57 and 1.72% for O<sub>2</sub> and CO<sub>2</sub>, respectively.

Storage of a mixture of CH<sub>4</sub> and 11.91% CO<sub>2</sub>-O<sub>2</sub> (Table I) resulted in no significant changes in the composition after 7 days at 20° but significant (P < 0.01) changes were observed after 14 days and the reproducibility of values between tubes was very poor. The change appeared to be an exchange of CH<sub>4</sub> for air. At -20° there were no significant changes after 14 days.

Through these trials a water peak of variable size appeared in all chromatograms. Attempts to quantitate the water proved futile as no reproducibility was obtained. However, as the analysis was being developed for use with rumen samples<sup>3</sup> and as reproducible results were obtained with water-saturated gas samples, the water content was considered a constant and was ignored. Using a different analytical system, it may be possible to estimate the water content.

The shelf life of the blood tubes will have an effect upon analytical results. Experience in drawing blood with these tubes indicates that the vacuum is reduced or lost with time. For this reason, only fresh tubes should be used for gas collection, tubes for each sample group should have the same lot number and blank values should be estimated with each analysis.

The data presented demonstrate the feasibility of using evacuated blood

<sup>\*</sup> Chromatograms of  $N_2 + O_2$  gave overlapping peaks. The  $O_2$  peak follows the  $N_2.$ 

<sup>\*\*</sup> One lot of tubes assayed in 1972 contained 0.5-1.0% CO.

collecting tubes for collecting, transporting, and briefly storing gas samples for analysis. Reproducibility of analysis was good and the amounts of gas originally present in the tubes could be reliably subtracted as an analytic blank. For gases of 5% (v/v), a CV of 1-2% was easily attained. Gases in smaller proportion (1-5%) were estimated with CV of less than 6%. Where this variability is acceptable, the blood tubes provide an inexpensive, convenient method for handling gas samples, particularly when large numbers of samples are to be collected at remote sites.

### CONCLUSION

Evacuated blood collecting tubes were tested for collection, transport and short term storage of gas samples for gas chromatography. Using a carrier-gas filled tube as a blank analysis of major components of air and rumen head-space gas could be achieved with CV < 5%.

## ACKNOWLEDGEMENT

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